

## Open Source QSAR Models For pKa Prediction Using Multiple Machine Learning Approaches

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### Background

- The logarithmic dissociation constant, pKa, strongly influences a chemical's pharmacokinetic and biochemical properties.
  - pKa reflects the ionization state of a chemical, which affects lipophilicity, solubility, protein binding, and the ability to cross the plasma membrane and the blood-brain barrier.
  - Thus, pKa affects absorption, distribution, metabolism, excretion, and toxicity (ADMET).
  - Chemicals with no charge at a physiological pH will passively cross the plasma membrane more easily than charged molecules and are therefore more likely to have biological activity than passively diffused charged chemicals.
- pKa is an important parameter for physiologically based pharmacokinetic (PBPK) modeling, in vitro to in vivo extrapolation (IVIVE), and predicting tissue:plasma partition coefficients.
- Commercial software tools such as ACD/Labs and ChemAxon predict the pKa of individual ionization sites independently of chemical class. However, current publicly available pKa models are limited to certain chemical classes.

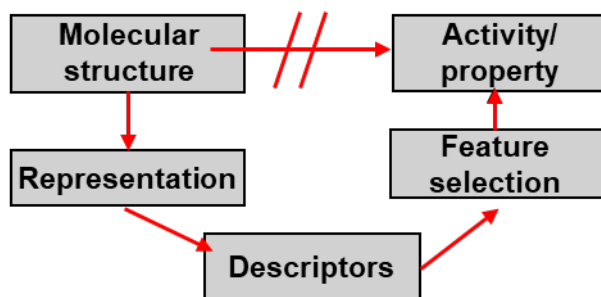
### Study Goals and Procedure

- Here we provide free, open-source, fast, and reliable options for predicting pKa for heterogeneous chemical classes.
- Modeling steps:
  - pKa values for 7912 chemicals in water were obtained from DataWarrior, a freely available software package.
  - Chemical structures were standardized for QSAR modeling (Mansouri et al. 2016).
  - Continuous molecular descriptors, binary fingerprints and fragment counts were generated using PaDEL.

- Several machine learning approaches were applied: deep neural networks (DNN), support vector machine (SVM), and extreme gradient boosting (XGB).
- Models were 5-fold cross-validated and evaluated against an external test set.
- The best models for each algorithm were compared to each other and to predictions from ACD/Labs and ChemAxon.

## QSAR Modeling

### Conceptual Basis



QSARs are based on the congenericity principle, which is the assumption that structurally similar compounds will have similar chemical properties.

QSARs can be fast and accurate but they depend on the quality of the data used.

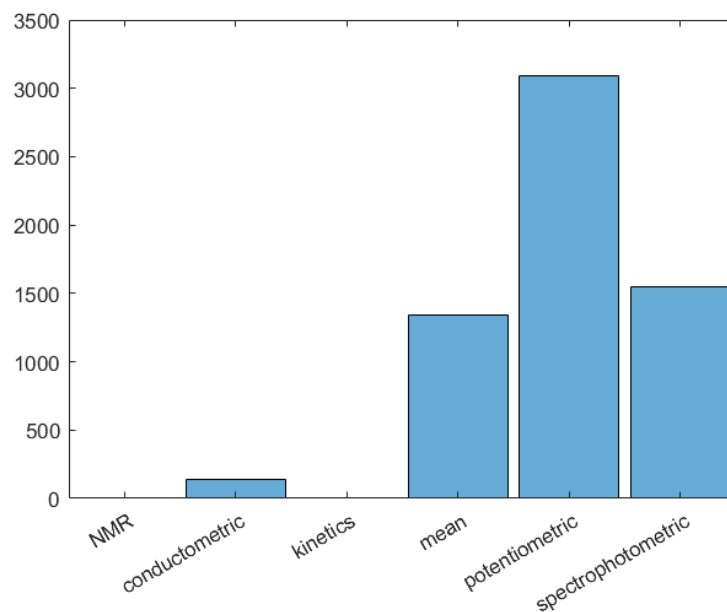
### General Steps to Develop a QSAR Model

- Curation of experimental data
- Standardization of the chemical structures
- Preparation of training and test sets
- Calculation of an initial set of descriptors
- Selection of a machine learning algorithm
- Variable selection technique
- Validation of the model's predictive ability
- Define the applicability domain
- Interpretation of the selected descriptors, if possible

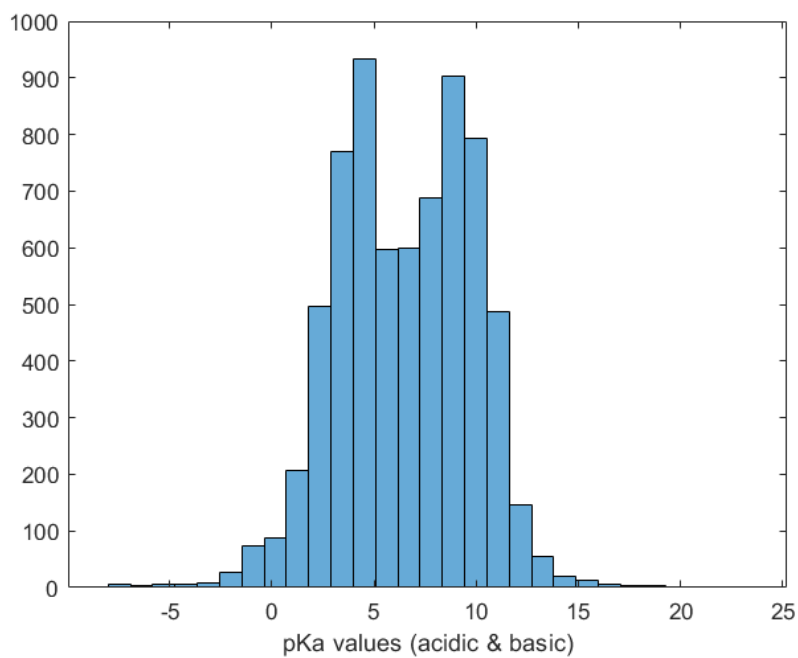
## pKa Data

- The pKa data were obtained from DataWarrior (<http://www.openmolecules.org/>) and included experimentally measured aqueous pKa values and associated SMILES strings for 7912 heterogenous chemicals.

### Methods for Measuring pKa Reported in DataWarrior



### Acidic and Basic pKa Values Reported in DataWarrior



## Data Preparation for Modeling

### Structure Standardization

#### Full dataset

7904 total valid structures

6245 unique QSAR-ready structures

#### Acidic dataset

3610 total valid structures

3260 unique QSAR-ready structures

#### Basic dataset

4294 total valid structures

3680 unique QSAR-ready structures

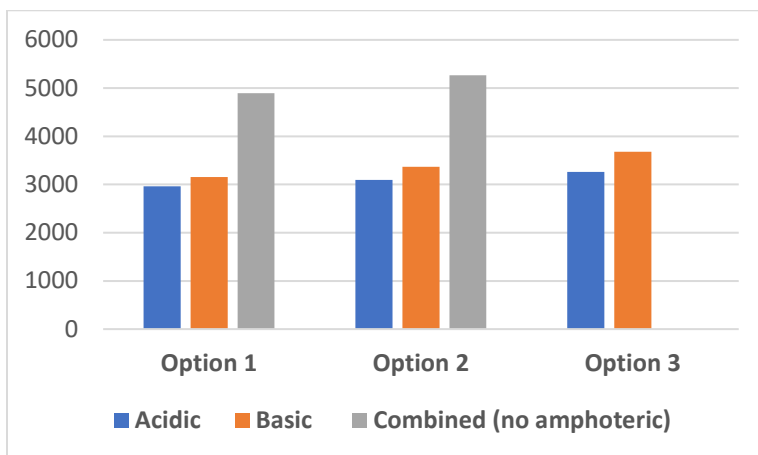


### Acidic and Basic Data Sets

The DataWarrior data set contained a high number of duplicates (1659) and amphoteric chemicals (chemicals with both an acidic and basic pKa). Data were processed in three different ways:

- Option 1: all duplicates removed
- Option 2: low variability duplicates averaged
- Option 3: all data included (strongest pKa rule)

QSAR-ready Structures in Each of the Data Options



### ***Training and Test Sets***

- For each data option, the structures were split into training (75%) and test (25%) sets.
- Training/test set splitting was performed semi-randomly to:
  - Keep similar distributions of pKa values
  - Keep similar distribution of acidic and basic pKas for combined datasets

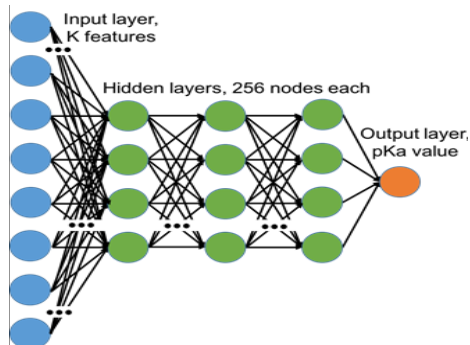
### ***Molecular Descriptors***

- The QSAR-ready structures were used to calculate molecular descriptors and generate binary fingerprints and fragment counts using PaDEL.
  - 1D and 2D continuous descriptors: 1444 descriptors
  - Binary fingerprints and counts: 9121 bits (CDK, Estate, MACCS, PubChem, Substructure, Klekota-Roth and 2D atom pairs)

## **Machine Learning Algorithms**

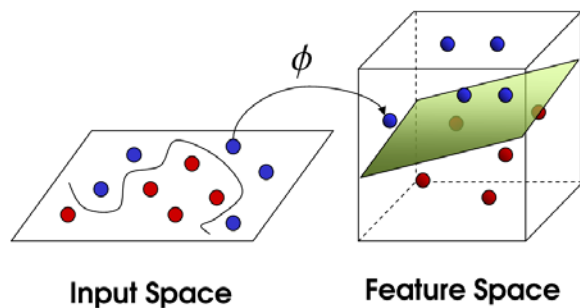
- All used tools and resulting models are free and open-source.

### ***Deep Neural Networks (DNN)***



- DNN maps features through a series of nonlinear functions that are linked in a combinatorial fashion to maximize model accuracy.
- Tensorflow and Keras packages were used to build a feed-forward DNN with 3 hidden layers of 256 nodes each.

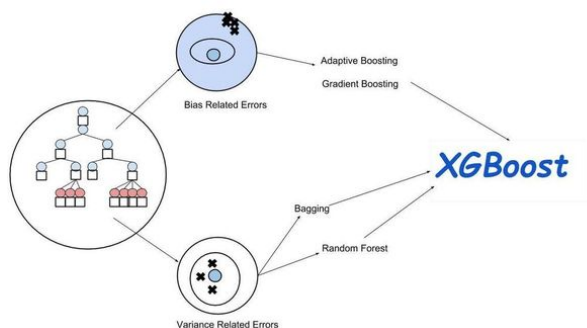
### ***Support Vector Machine (SVM)***



- SVM defines a non-linear decision boundary that optimally separates two classes.
- The free and open source package LibSVM3.1 was used for SVM implementation.

### ***Extreme Gradient Boosting (XGB)***

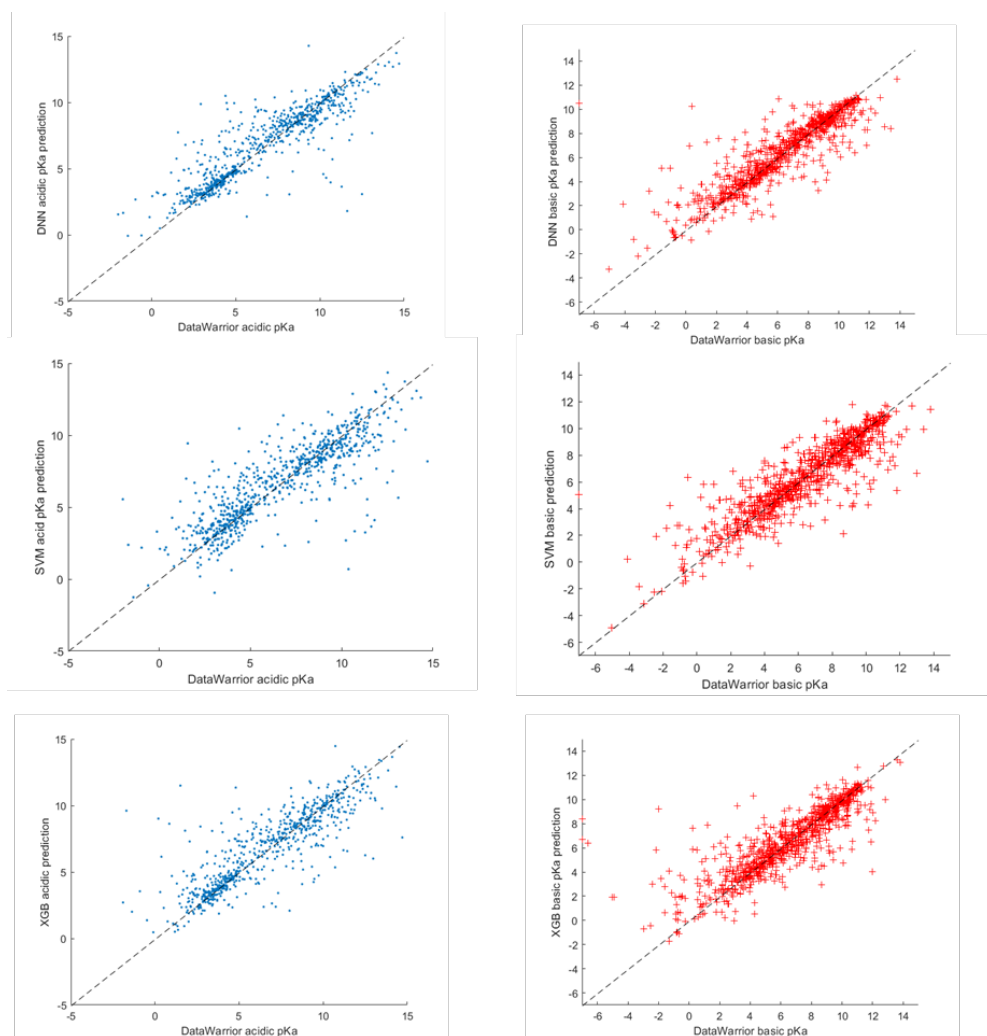
- XGB is used for regression and classification problems. The outputs of an ensemble of weak prediction models, typically decision trees, are combined to yield a final prediction.
- The R package caret was used to implement XGB.



## Model Performance

- The pKa dataset was divided into acidic and basic pKa datasets, which were modeled separately.
- Models were assessed using root mean-squared error (RMSE) and the coefficient of determination ( $R^2$ ). Test set results are reported below.

Algorithm	Best Acidic Model RMSE	Best Acidic Model $R^2$	Best Basic Model RMSE	Best Basic Model $R^2$
DNN	1.51	0.80	1.57	0.77
SVM	1.80	0.72	1.53	0.78
XGB	1.82	0.71	1.90	0.67



## Benchmark with Commercial Tools

### *Concordance Between the Commercial Tools and DataWarrior*

To be able to use the predictions from the commercial tools as a benchmark to our models, we first needed to assess the concordance of their predictions with DataWarrior.

#### **DataWarrior Acidic Dataset (3260 Chemicals)**

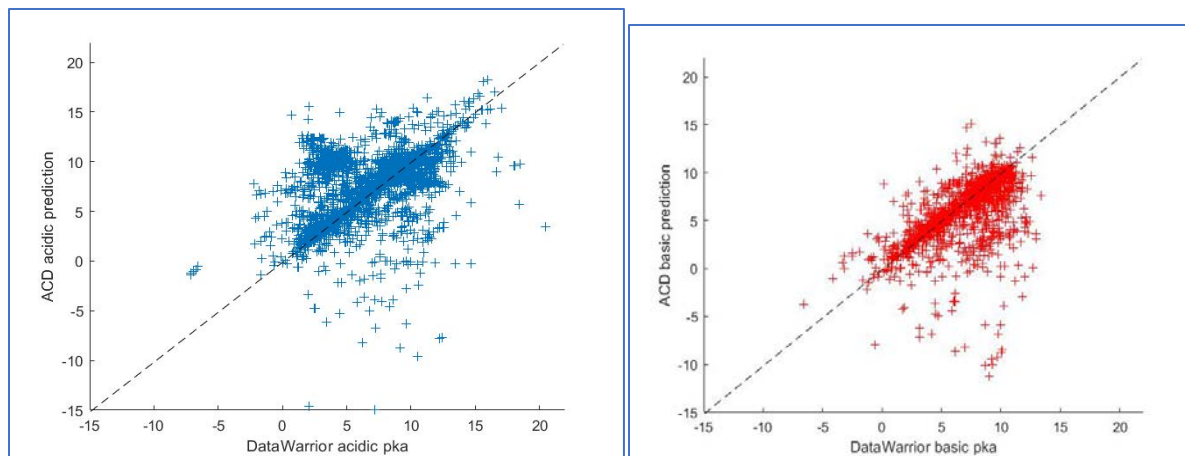
	<b>ACD/Labs</b>	<b>ChemAxon</b>
<b>Predicted chemicals</b>	3145	3206
<b>R<sup>2</sup></b>	-0.21	-0.11
<b>RMSE</b>	3.72	3.52

#### **DataWarrior Basic Dataset (3680 Chemicals)**

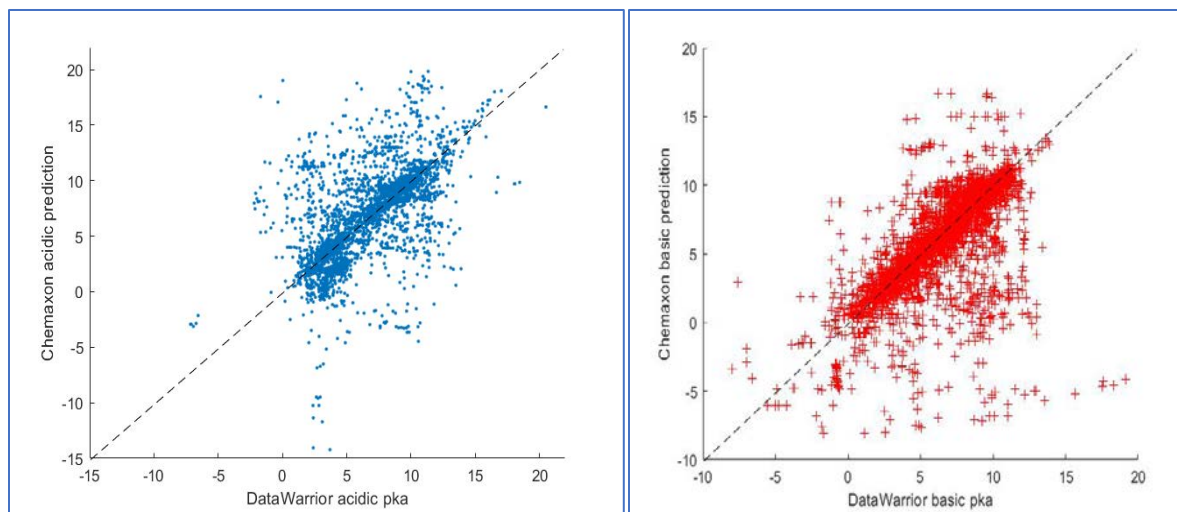
	<b>ACD/Labs</b>	<b>ChemAxon</b>
<b>Predicted chemicals</b>	1618	3649
<b>R<sup>2</sup></b>	-0.05	0.23
<b>RMSE</b>	3.00	2.79



### ACD/Labs pKa Predictor



### ChemAxon pKa Predictor



### External Set Prediction and Model Concordance

- A set of 8904 QSAR-ready structures (non-overlapping with DataWarrior) from the TSCA-actives list (<https://comptox.epa.gov/dashboard>) was used as benchmark to compare the predictions of the models from this work and the commercial tools.
- For this analysis, the SVM model was implemented in OPERA (<https://github.com/kmansouri/OPERA>) (Mansouri et al. 2018).

**Comparison of All Models for the Acidic pKa Predictions**

<b>Predictions</b>	<b>ACD/Labs - R<sup>2</sup></b>	<b>ACD/Labs - RMSE</b>	<b>ChemAxon - R<sup>2</sup></b>	<b>ChemAxon - RMSE</b>	<b>SVM - R<sup>2</sup></b>	<b>SVM - RMSE</b>	<b>DNN - R<sup>2</sup></b>	<b>DNN - RMSE</b>	<b>XGB - R<sup>2</sup></b>	<b>XGB - RMSE</b>
<b>ACD/Labs</b>	1	0	0.47	4.48	0.60	3.46	0.34	4.95	0.23	5.36
<b>ChemAxon</b>	0.60	4.48	1	0	0.52	4.55	0.45	5.41	0.30	6.09
<b>SVM</b>	*	*	*	*	1	0	0.51	2.09	0.44	2.27
<b>DNN</b>	*	*	*	*	0.74	2.09	1	0	0.51	2.39
<b>XGB</b>	*	*	*	*	0.43	2.27	0.15	2.39	1	0

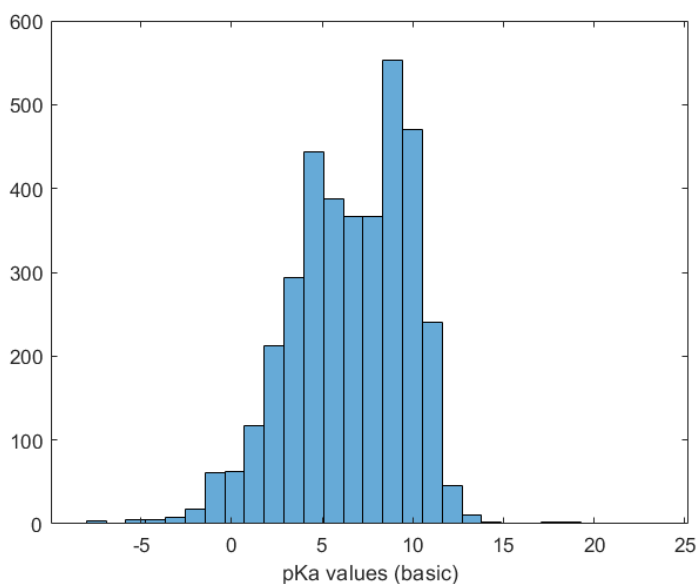
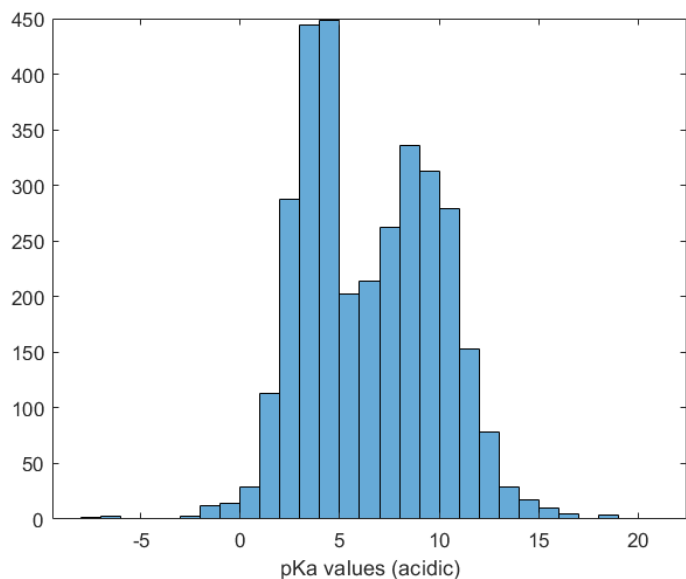
**Comparison of All Models for Basic pKa Predictions**

<b>Predictions</b>	<b>ACD/Labs - R<sup>2</sup></b>	<b>ACD/Labs - RMSE</b>	<b>ChemAxon - R<sup>2</sup></b>	<b>ChemAxon - RMSE</b>	<b>SVM - R<sup>2</sup></b>	<b>SVM - RMSE</b>	<b>DNN - R<sup>2</sup></b>	<b>DNN - RMSE</b>	<b>XGB - R<sup>2</sup></b>	<b>XGB - RMSE</b>
<b>ACD/Labs</b>	1	0	0.48	2.88	-0.14	4.57	-0.62	5.67	-0.80	5.99
<b>ChemAxon</b>	0.61	2.88	1	0	0.02	5.66	-2.62	9.77	-2.36	9.42
<b>SVM</b>	*	*	*	*	1	0	-0.90	3.41	0.37	1.97
<b>DNN</b>	*	*	*	*	0.15	3.41	1	0	0.35	2.99
<b>XGB</b>	*	*	*	*	0.28	1.97	-0.49	2.99	1	0

\* Our models are not used as reference to evaluate ChemAxon and ACD/Labs predictions.

## Range of Predictions and Limitations

### DataWarrior Acidic and Basic Datasets



- pKa predictions generated by our models range between about -5 and 15 for both the acidic and basic datasets.
- The narrow prediction range of our models (compared to the two commercial tools) is certainly linked to DataWarrior data that has the same range as shown by the distribution of its acidic and basic pKa values (histograms above).
- The different ranges in pKa predictions may also explain why:

- The disagreement between our models and the commercial models on the benchmark dataset (TSCA-actives) is higher for the basic pKa predictions. This is particularly noticeable with ChemAxon, which generated a high number of predictions of pKas lower than -5 for the basic data set.
- For the TSCA-actives list, the divergence between ACD/Labs and ChemAxon is higher for the basic pKa predictions compared to the acidic pKa predictions. Interestingly, this is the opposite of what occurred for the DataWarrior dataset.
- The predictions of our models can be considered more accurate in the range of -5 to 15 for both the acidic and basic pKas.

## Summary and Next Steps

- An automated QSAR data preparation workflow was applied to a public data set of 7912 chemicals, creating three data subsets, Acidic, Basic and Combined. Model performance was evaluated using all data subsets with the DNN, SVM, and XGB algorithms.
- The best models were compared and benchmarked with two commercial predictors showing different levels of concordance.
- The models and source codes will be available for download and use.
- This modeling effort will help provide predicted pKa values for all ionizable chemicals in the EPA DSSTox database.
  - Predictions will be available on the EPA's CompTox Chemistry Dashboard (<https://comptox.epa.gov>)
  - Predictions will also be used by the NICEATM's Integrated Chemical Environment (ICE) Dashboard (<https://ice.ntp.niehs.nih.gov/>) in various pharmacokinetic calculations.

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## **References**

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